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View Notes - [note_taking_guide_702](#) from CHEM 303 at Wheaton College. Note Taking Guide: Episode Name Example Problem: Find the % by mass of oxygen in water. $(16/18) * 100 = 88.9\%$ Percentage by mass

Physics 702: Torque and Rotational Equilibrium | Georgia ...

At the completion of this episode's lesson(s), you should be able to: • Compare and contrast translational and rotational motion. • Distinguish between stable and unstable rotational equilibrium. • Define rotational inertia and list factors affecting it. • Define torque and lever arm.

Note Taking Guide Episode 605 Answer Key

Chemistry 702: Percentage Composition and Empirical Formulas Instructions Before viewing an episode, download and print the note-taking guides, worksheets, and lab data sheets for that episode, keeping the printed sheets in order by page number.

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Example Problem: Find the % by mass of oxygen in water. Percentage by mass of element in a compound = $(\text{mass of element in 1 mol of compound} \div \text{molar mass of compound}) \times 100\%$

Example Problem: Find the % by mass of oxygen in water.

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directions are _ or _ . direction is _ direction is _ Problem set #1:

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